134. Catalytic Intramolecular Palladium-Ene Reactions

Preliminary Communication¹)

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Pd(dba)₂[dba = dibenzylideneacetone]/PPh₃-or Pd(PPh₃)₄-catalyzed cyclizations of acetoxy-dienes $2 \rightarrow 3$ and $10 \rightarrow 11$ gave 1-vinyl-2-methylidene-substituted cyclopentanes and cyclohexanes in high yield, consistent with a palladium-ene/ β -elimination mechanism ($D \rightarrow E \rightarrow F$, Scheme 2). The efficient and highly stereoselective cyclizations $4 \rightarrow 7$ and $8 \rightarrow 9$ illustrate intramolecular allylpalladium insertions into 1,2-dialkyl-, trialkyl-, and cyclic alkenes followed by elimination of the exocyclic β -H giving 1,2-divinylcyclopentanes. These new olefin insertions proceed faster in AcOH (compared to THF) and occur preferentially *cis* relative to the Pd ($13 \rightarrow 14 \rightarrow 15$).



In conjunction with our previous studies of the magnesium-ene cyclization [1] (*Scheme 1*), we envisaged the extension of this concept to catalytic intramolecular palladium-ene reactions²) (*Scheme 2*).



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²) Review articles on Pd-mediated reactions: [2]. For Pd^{II}-catalyzed cyclizations of enynes, see [3].

Whereas norbornadiene [4], norbornene [4], and 1,3-dienes [5] were reported to insert into stoichiometric amounts of allylpalladium complexes, simple olefins (*e.g.* styrene, cyclohexene, 1,4-cyclohexadiene, and 1,5-cyclooctadiene) did not undergo this reaction under similar conditions [4a]. Nevertheless, we assumed the intramolecular ene process $D \rightarrow E$ to be entropically facilitated and a subsequent irreversible β -elimination $E \rightarrow F$ to withdraw the ene product E from the equilibrium $D \leftrightarrows E$.

In-situ preparation of the olefinic allylpalladium intermediates C could be accomplished by oxidative addition of Pd° complexes to allyl-acetates (A, B) [2]. The required 1-acetoxy-2,7(8)-dienes A were readily obtained in 68 to 95% yield, predominantly as their (E)-isomers, via Pd(PPh₃)₄-catalyzed alkylation of 1-acetoxy-4-chloro-2-butene [6] or, preferably, (4-acetoxy-2-butenyl)methyl carbonate³)⁴) with a 3(4)-alkenyl-1,1-disulfone³), or with 3-alkenyl-malonates as exemplified by the transformation $1a^4 \rightarrow 2^4$) (Scheme 3).



Heating diene **2a** with bis(dibenzylideneacetone)palladium [Pd(dba)₂ (= bis[η -1,5-diphenyl-1,4-pentadien-3-one]palladium); 0.07 equiv.] and PPh₃ (0.2 equiv.) in THF at +70° for 2 h gave the expected cyclized 1,4-diene **3a**⁴) in 83% yield (*Scheme 3, Table 1*). Even more conveniently, product **3a** was obtained in one operation from disulfone **1a** (*Entry 2*) by treatment with (4-acetoxy-2-butenyl)methyl carbonate (1.0 equiv.), Pd(dba)₂ (0.07 equiv.), and PPh₃ (0.2 equiv.) in THF at +20° (2 h) and then at +75° (2 h). Accordingly, bonds C(4)–C(5) and C(1)–C(2) of cyclopentane **3a** may be efficiently formed by this simple combined alkylation/cyclization procedure.

Entry	Series	Y	Solvent	mol-equiv. PPh ₃	Reaction temp. [°C] (Reaction time)	Yield [%] of 3
1	a	$(SO_2 - p - MePh)_2$	THF	0.2	+70° (2 h)	82
2 ^a)	а	$(SO_2 - p - MePh)_2$	THF	0.2	'l-pot'a)	76 ^a)
3	b	(COOMe) ₂	THF	0.2	$+80^{\circ}$ (40 h)	20
4	b	$(COOMe)_2$	MeOH	0.2	+80° (8 h)	65
5	b	(COOMe) ₂	AcOH	0.2	$+80^{\circ}$ (1.5 h)	77
6	b	(COOMe) ₂	AcOH	-	+80° (3 h)	
7	c	$(COO)_2 CMe_2$	AcOH	0.2	+80° (1.5 h)	75

Table 1. $Pd(dba)_2/PPh_3$ -Catalyzed Cyclizations $2 \rightarrow 3$

^a) Pd(dba)₂ (15 mg) and PPh₃ (20 mg) were added to 1a (140 mg) in THF (4 ml). Stirring the mixture under Ar at r.t. for 2 h and then at 75° for 2 h, followed by aq. workup and flash chromatography, gave pure 3a (122 mg, 76% yield from 1a; m.p. 166–167°).

³) Prepared from (Z)-1-acetoxy-2-buten-4-ol [7] with ClCOOMe (1.1 mol-equiv.), pyridine (1.5 mol-equiv.), CH₂Cl₂, 0° → +20°, < 95% yield. 1a was prepared by mono-alkylation of bis(tolylsulfonyl)methane [8].</p>

⁴) All isolated intermediates and products were characterized by IR, ¹H-NMR (360 MHz), ¹³C-NMR, and MS.

Solvent effects significantly influence this novel ene process as illustrated by the cyclization of malonate **2b**. Whereas no reaction took place in toluene, CH_2Cl_2 , or DMF, the rate and yield increased on proceeding from THF to MeOH to AcOH (*Entries 3–5*). Interestingly, the presence of the phosphine turned out to be indispensable (*Entry 6*).

In striking contrast to 8-alkyl-substituted 2,7-dienylmagnesium halides, which did not cyclize [9], the allylpalladium unit of 5 inserted readily into a terminally mono- and even dimethyl-substituted olefinic bond ($5 \rightarrow 6$; Scheme 4, Table 2).



Table 2. $Pd(dba)_2/PPh_3$ -Catalyzed Cyclizations $4 \rightarrow 7$

Entry	Series	R	Y	Solvent	Reaction temp. [°C] (Reaction time)	Yield [%] of 7
8	a	Н	$(SO_2 - p - MePh)_2$	THF	+25° (15 h)	80
9 ^a)	а	Н	$(SO_2 - p - MePh)_2$	AcOH	+75° (1.5 h)	91
10	b	Me	$(SO_2 - p - MePh)_2$	THF	+85° (40 h)	40
<u>11</u>	b	Me	$(SO_2 - p - MePh)_2$	AcOH	+75° (1.5 h)	71

^a) Pd(dba)₂ (8 mg) and PPh₃ (11 mg) were added to 4a (98 mg) in AcOH (3 ml). Heating the mixture at +75-80° for 1.5 h, addition of sat. aq. Na₂CO₃, workup, and flash chromatography furnished pure 7a (86 mg, 91% yield; m.p. 136-137°).

Thus, Pd°-catalyzed cyclizations of 1-acetoxy-2,7-dienes $4a^4$) and $4b^4$) gave in each case a single 1,5-diene product $7a^4$)⁵) and $7b^4$)⁵), respectively. It follows that the ene-step $5 \rightarrow 6$ is highly stereoselective in both cases and that the intermediates 6 eliminate the exocyclic H_b preferentially over H_a, in agreement with the conformational constraints of a *syn-β*-elimination process [2]. Again, the cyclizations $5 \rightarrow 6$ proceeded significantly faster in AcOH as compared to THF without change of the stereochemical outcome. In contrast, attempts to effect a Pd°-catalyzed cyclization of 4a in THF, while trapping the generated AcOH by finely powdered Na₂CO₃, gave only unchanged 4a.

As expected, $Pd(PPh_3)_4$ (0.07 equiv.) turned out to be an equally efficient catalyst for intramolecular palladium-ene reactions (*Scheme 5*).

Thus, 1-acetoxy-2,7-diene 8^4) containing a cyclic 'enophile' unit furnished stereoselectively the bicyclic product 9^4)⁵) (m.p. 176–177°) in 86% yield. Similar conversion of the 1-acetoxy-2,8-diene 10^4) to 11^4) illustrates the feasibility of this method for 6-memberedring formation.

⁵) The relative configuration of this single stereoisomer has not yet been established.



The palladium-ene unit may also be part of a ring as shown by the smooth $Pd(PPh_3)_4$ catalyzed cyclization of the *trans*-acetoxydiene **13**^{4,6}) which was complete after 1.1 h to give the *cis*-fused hydrindene **15**⁴) in 84% yield (*Scheme 6*).



Under similar conditions, the *cis*-substituted acetoxydiene 16^{4})⁶) reacted slower, affording after 4 h the identical *cis*-product 15^{4}) in only 55% yield. These results indicate that the olefin inserts into the allylpalladium unit preferentially *cis* relative to the Pd, consistent with the intermediacy of 14^{6}). However, in the epimeric complex 17^{6}), coordination of the Pd with the *trans*-disposed enophile is excluded. Therefore, the slower conversion of 16 to 15 may imply that 17 isomerizes to 14 (*e.g. via* the corresponding π -allyl complex), or that 17 undergoes a relatively slow 'enophile'-insertion *trans* to the Pd.

⁶) Preparation of **12**: [10]. The relative configurations of intermediates **13**, **14**, **16**, and **17** were assigned on the basis that oxidative additions of Pd^e to allyl chlorides and allylacetates proceeds with inversion and that the resulting allylpalladium complexes are substituted by malonate anions *trans* relative to the Pd¹¹ [2].

In summary, we have shown here that catalytic intramolecular palladium-ene reactions are simple to carry out, compatible with various functional groups7) as well as applicable to 1,2-dialkyl-, trialkyl-, and cycloalkenyl enophiles, thus, complementing advantageously the analogous magnesium-ene process. Further extensions and applications of this novel methodology are presently under investigation in our laboratories.

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⁷⁾ For the smooth hydrogenolytic removal (Na, Hg) of geminal disulfones without reduction of isolated olefinic bonds, see [11].